# ent-9(8 $\rightarrow 15 \alpha$ H)abeo-17-Norkaur-8(14)-en-16-one 

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#### Abstract

C}_{19} \mathrm{H}_{28} \mathrm{O}\), orthorhombic, $P 2_{2} 2_{1} 2_{1}, a=$ 23.69 (4), $b=8.058$ (12), $c=8.132$ (12) $\AA, Z=4$. The two six-membered rings are in the chair form, and the five-membered ring is in an envelope conformation. All the rings in the moleculre are somewhat distorted. Repulsion between the 1,3 diaxial methyl groups attached to ring $A$ results in rotation of the methyl H atoms from the usual staggered position.


Introduction. Two of the authors (MO and EF) carried out a synthesis of ent-9 ( $8 \rightarrow 15 \alpha \mathrm{H}$ ) abeo-kaurane (I). In this synthesis, ent-17-norkauran-16-one (II) was allowed to react with thallium trinitrate to afford a rearranged product which was assigned structure III on the basis of spectral and chemical evidence (Fujita \& Ochiai, 1976). The present X-ray analysis of compound III provides a confirmation of the structure and
also stereochemical evidence for the large strain expected in the structure.

(1)

(II)

(iII)

Intensity data were collected on a Rigaku automatic four-circle diffractometer using $\mathrm{Cu} K \alpha$ radiation. The structure was solved by trial and error methods based on Patterson maps and $E$ maps derived by use of the program MULTAN (Germain, Main \& Woolfson, 1971). Successive cycles of anisotropic block-diagonal least-squares refinement gave an $R$ value of 0.043 for 1571 observed reflexions. $H$ atoms were located from

Table 1. Fractional atomic coordinates and anisotropic thermal parameters $\left(\times 10^{4}\right)$ of the non-hydrogen atoms with estimated standard deviations in parentheses

The temperature factor expression is $T=\exp \left[-\left(h^{2} \beta_{11}+k^{2} \beta_{22}+l^{2} \beta_{33}+2 h k \beta_{12}+2 h l \beta_{13}+2 k l \beta_{23}\right)\right]$.

|  | $x$ | $y$ | $z$ | $\beta_{11}$ | $\beta_{22}$ | $\beta_{33}$ | $\beta_{12}$ | $\beta_{13}$ | $\beta_{23}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C(1) | 1630 (2) | 1034 (6) | 4036 (6) | 23 (1) | 121 (8) | 125 (8) | 3 (3) | -7 (3) | 21 (8) |
| C (2) | 1883 (2) | 2276 (7) | 5264 (6) | 34 (1) | 164 (9) | 121 (8) | 7 (3) | -17(3) | 23 (9) |
| C(3) | 1527(2) | 3865 (6) | 5297 (6) | 26 (1) | 163 (9) | 136 (8) | -4 (3) | 2 (3) | 18 (9) |
| $\mathrm{C}(4)$ | 1455 (2) | 4698 (6) | 3613 (6) | 15 (1) | 144 (8) | 156 (8) | -6(2) | -5 (2) | -1 (8) |
| C(5) | 1258 (2) | 3377 (5) | 2318 (5) | 12 (1) | 104 (7) | 133 (7) | -5 (2) | -5 (2) | 21 (7) |
| C(6) | 1158 (2) | 4168 (6) | 609 (6) | 20 (1) | 125 (8) | 159 (8) | -7(2) | -18(3) | 39 (8) |
| C(7) | 575 (2) | 3654 (6) | -132(6) | 18 (1) | 132 (8) | 190 (9) | 2 (2) | -22(3) | 14 (8) |
| C(8) | 552 (2) | 1821 (5) | -441 (6) | 11 (1) | 122 (7) | 176 (9) | 1 (2) | -11(2) | 21 (8) |
| C(9) | 1315 (2) | 337 (5) | 1183 (5) | 11 (1) | 111 (7) | 126 (7) | 0 (2) | -2 (2) | 9 (7) |
| C(10) | 1606 (2) | 1727 (5) | 2263 (5) | 11 (1) | 119 (7) | 116 (7) | 0 (2) | -1 (2) | 18 (7) |
| C(11) | 1583 (2) | 43 (6) | -534 (6) | 14 (1) | 165 (9) | 128 (8) | 3 (2) | 4 (2) | -14 (8) |
| $\mathrm{C}(12)$ | 1256 (2) | -1201 (6) | -1614 (6) | 18 (1) | 160 (9) | 152 (8) | 7 (3) | 1 (2) | -8 (8) |
| C(13) | 612 (2) | -816(6) | -1644 (6) | 17 (1) | 150 (9) | 165 (9) | -6(3) | -11(3) | -12(8) |
| C(14) | 516 (2) | 1028 (6) | -1871 (6) | 16(1) | 163 (9) | 159 (9) | -1(3) | -13(2) | 16 (8) |
| C(15) | 659 (2) | 598 (5) | 951 (6) | 12 (1) | 112 (7) | 141 (8) | -3 (2) | 1 (2) | 10 (7) |
| C(16) | 438 (2) | -981 (5) | 148 (6) | 12 (1) | 119 (7) | 190 (9) | 0 (2) | -6 (2) | 2 (8) |
| $\mathrm{O}(17)$ | 183 (1) | -2095 (4) | 789 (5) | 19 (1) | 147 (6) | 245 (7) | -14(2) | 2 (2) | 16 (6) |
| C(18) | 981 (2) | 5996 (6) | 3814 (7) | 22 (1) | 148 (9) | 261 (12) | 4 (3) | -12(3) | -39(10) |
| C(19) | 1998 (2) | 5649 (6) | 3134 (7) | 21 (1) | 180 (10) | 214 (11) | -25(3) | -12(3) | -10(10) |
| C(20) | 2220 (2) | 1965 (7) | 1659 (6) | 11 (1) | 204 (10) | 185 (9) | -5 (2) | 1 (2) | -13 (9) |

the difference synthesis and included in the refinement assuming an isotropic temperature factor of $3.2 \AA^{2}$. The final structural parameters are listed in Tables 1 and 2.* A perspective view of the molecule and the crystal packing are shown in Figs. 1 and 2 respectively. The bond lengths and angles for the non-hydrogen atoms are given in Fig. 3.
*A list of structure factors has been deposited with the British Library Lending Division as Supplementary Publication No. SUP 32131 ( 7 pp .). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 13 White Friars, Chester CH1 1NZ, England.

Table 2. Fractional atomic coordinates $\left(\times 10^{3}\right)$ of the hydrogen atoms with estimated standard deviations in parentheses

|  |  |  | $y$ |
| :--- | ---: | ---: | ---: |
|  | $x$ | $z$ |  |
| $\mathrm{H}(1 A)$ | $123(2)$ | $73(6)$ | $434(6)$ |
| $\mathrm{H}(1 B)$ | $188(2)$ | $-9(6)$ | $404(6)$ |
| $\mathrm{H}(2 A)$ | $235(2)$ | $249(6)$ | $504(6)$ |
| $\mathrm{H}(2 B)$ | $190(2)$ | $173(6)$ | $644(6)$ |
| $\mathrm{H}(3 A)$ | $170(2)$ | $474(6)$ | $614(6)$ |
| $\mathrm{H}(3 B)$ | $109(2)$ | $361(6)$ | $578(6)$ |
| $\mathrm{H}(5)$ | $86(2)$ | $305(6)$ | $279(6)$ |
| $\mathrm{H}(6 A)$ | $151(2)$ | $389(6)$ | $-26(6)$ |
| $\mathrm{H}(6 B)$ | $118(2)$ | $551(6)$ | $77(6)$ |
| $\mathrm{H}(7 A)$ | $50(2)$ | $429(6)$ | $-124(6)$ |
| $\mathrm{H}(7 B)$ | $21(2)$ | $406(6)$ | $62(6)$ |
| $\mathrm{H}(9)$ | $138(2)$ | $-78(6)$ | $191(6)$ |
| $\mathrm{H}(11 A)$ | $199(2)$ | $-45(6)$ | $-37(6)$ |
| $\mathrm{H}(11 B)$ | $163(2)$ | $114(6)$ | $-107(6)$ |
| $\mathrm{H}(12 A)$ | $133(2)$ | $-239(6)$ | $-129(6)$ |
| $\mathrm{H}(12 B)$ | $140(2)$ | $-115(6)$ | $-273(6)$ |
| $\mathrm{H}(13)$ | $41(2)$ | $-156(6)$ | $-248(6)$ |
| $\mathrm{H}(14)$ | $48(2)$ | $149(6)$ | $-309(6)$ |
| $\mathrm{H}(15)$ | $46(2)$ | $88(6)$ | $209(6)$ |
| $\mathrm{H}(18 A)$ | $88(2)$ | $677(6)$ | $290(6)$ |
| $\mathrm{H}(18 B)$ | $106(2)$ | $678(6)$ | $476(6)$ |
| $\mathrm{H}(8 C)$ | $58(2)$ | $548(6)$ | $427(6)$ |
| $\mathrm{H}(19 A)$ | $200(2)$ | $593(6)$ | $189(6)$ |
| $\mathrm{H}(19 B)$ | $235(2)$ | $507(7)$ | $354(6)$ |
| $\mathrm{H}(19 C)$ | $201(2)$ | $666(6)$ | $362(6)$ |
| $\mathrm{H}(20 A)$ | $226(2)$ | $249(6)$ | $51(6)$ |
| $\mathrm{H}(20 B)$ | $239(2)$ | $86(6)$ | $164(6)$ |
| $\mathrm{H}(20 C)$ | $244(2)$ | $280(6)$ | $237(6)$ |



Fig. 1. Perspective view of the molecule.

Discussion. Six-membered rings $A$ and $C$ are in a distorted chair form. Torsional angles for the ring $A$ (49.3-59.8 ${ }^{\circ}$ ) and for the ring $C\left(45 \cdot 9-77 \cdot 1^{\circ}\right)$ deviate largely from the idealized value for the chair form. Seven-membered ring $B$ is also distorted owing to repulsion between the H atoms on $\mathrm{C}(5)$ and $\mathrm{C}(15)$. The internal bond angles at $\mathrm{C}(5), \mathrm{C}(9)$ and $\mathrm{C}(10)$ are significantly larger than the normal tetrahedral angle.


Fig. 2. Crystal packing. Axes for the unit cell are $a \downarrow, b \rightarrow$, and $c$ is out of the page.

(b)

Fig. 3. (a) Bond lengths (standard deviations range from 0.006 to $0.007 \AA$ ). (b) Bond angles (standard deviations range from 0.4 to $0.5^{\circ}$ ).

Table 3. Least-squares planes and atomic deviations
The planes are defined by $A x+B y+C z+D=0$ where $x, y$ and $z$ are in $\AA$ along the axes $a, b$ and $c$. An asterisk indicates an atom not used in the plane calculation.

| $A=-0.8634$ |  | $A=-0.9905$ |  |
| :---: | :---: | :---: | :---: |
| $B=0$ |  | $B=-0.0654$ |  |
| $C=-0.2037$ |  | $C=0.1208$ |  |
| $D=1.2841$ |  | $D=1.4961$ |  |
| Atom deviations ( $\AA$ ) |  | Atom deviations ( $\AA$ ) |  |
| C(7)* | 1.489 | C (7) | -0.059 |
| C(8)* | 0.905 | C(8) | 0.061 |
| $\mathrm{C}(13)$ | 0.001 | $\mathrm{C}(13)$ | -0.059 |
| $\mathrm{C}(14)^{*}$ | 0.920 | C(14) | 0.046 |
| C(15) | 0.001 | C(15) | 0.011 |
| C(16) | -0.002 | C(16)* | 0.534 |
| $\mathrm{O}(17)$ | 0.001 | $\mathrm{O}(17)^{*}$ | 1.255 |
| H(14)* | 1.372 | $\mathrm{H}(14)^{*}$ | -0.008 |

The single bond $C(7)-C(8)(1.499 \AA)$ is shortened by resonance with the adjacent double bond $\mathrm{C}(8)-\mathrm{C}(14)$. Five-membered ring $D$ has an envelope conformation. The $\mathrm{C}(16)$ atom deviates from the plane of the other four atoms. The least-squares plane through $\mathrm{C}(13)$, $\mathrm{C}(15), \mathrm{C}(16)$ and $\mathrm{O}(17)$ makes a dihedral angle of $33.8^{\circ}$ with that through $\mathrm{C}(7), \mathrm{C}(8), \mathrm{C}(13), \mathrm{C}(14)$ and $\mathrm{C}(15)$ (Table 3). The strain in ring $D$ is evidenced by the fact that the internal bond angles at $\mathrm{C}(8), \mathrm{C}(14)$ and $\mathrm{C}(16)$ are much smaller than the normal $s p^{2}$-hybridization angle ( $120^{\circ}$ ). Repulsion between the 1,3 diaxial methyl groups attached to ring $A$ causes the significantly greater bond angles $114.5^{\circ}$ for $C(5)-C(4)-C(19)$ and
$113.5^{\circ}$ for $\mathrm{C}(5)-\mathrm{C}(10)-\mathrm{C}(20)$, resulting in the $\mathrm{C}(19)-\mathrm{C}(20)$ distance of $3.24 \AA$. A similar structural feature has been observed in ( - )-kaur-15-en-19-al (Karle, 1972) and in phyllocladan- 15 -yl bromoacetate (Brown \& Hall, 1976). In the present case, the three methyl H atoms on $\mathrm{C}(19)$ are rotated about the $\mathrm{C}(4)-$ $\mathrm{C}(19)$ bond from the staggered position, while the methyl H atoms on $\mathrm{C}(20)$ are in the usual staggered position. The shortest $\mathrm{H}-\mathrm{H}$ approach between these methyl groups is $2.08 \AA$.

There are no intermolecular approaches likely to influence the molecular structure.

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## References

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